

- (c) Write down the types of electroanalytical method and the basic difference between the Thermal Analysis and Thermo Gravimetric Analysis. 3+5
- (d) (ii) Write down the application of Cyclic Voltammetry (CV). (ii) Why Ferrocene is used as standard in CV? (iii) Draw the Cyclic Voltammogram of Reversible, Irreversible and Quasi-Reversible redox reaction. 3+2+3

.....

**Internal Assessment-10**

Total Pages -02

PKC/PG/IVS/CEM-403/23

**2023**

**M.Sc.**

**4<sup>th</sup> Semester Examination**

**CHEMISTRY**

**PAPER – CEM-403 (Inorganic Special)**

**Full Marks : 50**

**Time : 2 Hours**

**(CEM 403-Advanced Inorganic Chemistry-II)**

**1. Answer any *four* bits:** 2×4 = 8

- (a) What is the importance of Electroanalytical chemistry?
- (b) What is substitution reaction?
- (c) Define Fick's law.
- (d) What is liquid junction potential?
- (e) What is polarography?
- (f) Write down the difference between DTA and DSC?

**2. Answer any *four* bits:** 4×4 = 16

- (a) Discuss Transition state theory of inorganic reaction mechanism.
- (b) Write down the difference between conductometric and potentiometric titration?
- (c) Define Complementary and Non-Complementary Reactions.
- (d) Write down advantages and disadvantages of DME?
- (e) Explain the nature of thermogram (TGA) of Calcium oxalate and magnesium oxalate.
- (f) Write down the difference between Electrolytic cell and Electrochemical cell.

**3. Answer any *two* questions** 2×8 = 16

- (a) Discuss the associative and dissociative mechanism for octahedral complex and explain the Mechanism of Two Electron Transfer Reactions. 4+4
- (b) Write the basic principle of Ilkovic-Heyrovsky equation. Explain the significance of this equation. 4+4

- (c) Write down the types of electroanalytical method and the basic difference between the Thermal Analysis and Thermo Gravimetric Analysis. 3+5
- (d) (ii) Write down the application of Cyclic Voltammetry (CV). (ii) Why Ferrocene is used as standard in CV? (iii) Draw the Cyclic Voltammogram of Reversible, Irreversible and Quasi-Reversible redox reaction. 3+2+3

.....

**Internal Assessment-10**

Total Pages -02

PKC/PG/IVS/CEM-403/23

**2023**

**M.Sc.**

**4<sup>th</sup> Semester Examination**

**CHEMISTRY**

**PAPER – CEM-403 (Inorganic Special)**

**Full Marks : 50**

**Time : 2 Hours**

**(CEM 403-Advanced Inorganic Chemistry-II)**

**1. Answer any *four* bits:** 2×4 = 8

- (a) What is the importance of Electroanalytical chemistry?
- (b) What is substitution reaction?
- (c) Define Fick's law.
- (d) What is liquid junction potential?
- (e) What is polarography?
- (f) Write down the difference between DTA and DSC?

**2. Answer any *four* bits:** 4×4 = 16

- (a) Discuss Transition state theory of inorganic reaction mechanism.
- (b) Write down the difference between conductometric and potentiometric titration?
- (c) Define Complementary and Non-Complementary Reactions.
- (d) Write down advantages and disadvantages of DME?
- (e) Explain the nature of thermogram (TGA) of Calcium oxalate and magnesium oxalate.
- (f) Write down the difference between Electrolytic cell and Electrochemical cell.

**3. Answer any *two* questions** 2×8 = 16

- (a) Discuss the associative and dissociative mechanism for octahedral complex and explain the Mechanism of Two Electron Transfer Reactions. 4+4
- (b) Write the basic principle of Ilkovic-Heyrovsky equation. Explain the significance of this equation. 4+4

Page-01

**(Turn over)**