

**2018****M.Sc.****3<sup>rd</sup> Semester Examination****PHYSICS****PAPER – PHS-302 (Gr. – A + B)*****Full Marks : 50******Time : 2 Hours******(Molecular Spectroscopy and Laser Physics – PHS 302A)******Answer Q1 and any one from Q2 and Q3***

1. Answer any five bits: 5X2 = 10
- (a) Which one of the following molecules does not exhibit a rotational spectrum and why? H<sub>2</sub>, CO, HCl, HBr
- (b) In the rotational spectrum of <sup>12</sup>C<sup>16</sup>O absorption lines are observed at a constant separation of 3.84236 cm<sup>-1</sup>. Calculate angular velocity in radian per second in  $j = 1$  level.
- (c) Find the ratio of the rates of spontaneous and stimulated emissions at  $T=10^3$  K for visible radiation of frequency  $5 \times 10^{14}$  Hz.
- (d) What do you mean by Quality factor of a laser resonator?
- (e) Discuss the non existence of two level lasing systems?
- (f) State Frank-Condon principle for intensity distribution of lines in molecular spectra.
- (g) Draw the energy level diagram of Ruby laser.

***(Turn Over)***

(h) Why a spherical symmetric molecule is microwave inactive? Give an example of such molecule.

2. (a) Discuss rotational spectra of a diatomic molecule treated as a non-rigid rotator. (4)

(b) What do you mean by zero-point energy? And derive this for a diatomic molecule. (3)

(c) The fundamental and first overtone transition of  $^{14}\text{N}^{16}\text{O}$  are centered at  $1876.06\text{ cm}^{-1}$  and  $3724.20\text{ cm}^{-1}$  respectively. Evaluate the equilibrium vibration frequency and anharmonic constant. (3)

3. (a) A He-Ne laser has a coherence length of 10m. What is the coherence time? What is population inversion? (1+1)

(b) What is a four-level laser system? Obtaining the rate equations of each of the energy levels, find the expression of population inversion in the system. (1+5)

(c) Why four-level laser is more efficient than three-level laser? (2)

**(Nuclear Physics I – PHS 302B)**

**Answer Q1 and any one from Q2 and Q3**

1. Answer any five bits: 5X2 = 10

(a) What is Kurie plot? What is its significance?

(b) Present diagrammatically the mechanism of  $\alpha$ -decay.

**(Continued)**

(c) Explain the nuclear shape when its electric quadrupole moment (Q) becomes  $Q = 0$ ,  $Q > 0$  and  $Q < 0$ .

(d) Graphically show the energy spectra of  $\beta^+$  and  $\beta^-$  particles in  $\beta$ -decay.

(e) Write down the selection rules for allowed Gamow-Teller transition.

(f) Write down the principle of double focusing mass spectrograph.

(g) Using the semi-empirical mass formula show that the most stable isobar for a nucleus having odd  $A$  is given by  $Z = \frac{A}{0.015A^{2/3}+2}$

(h) Determine the stable nucleus that has radius is equal to 1/3 that  $^{189}\text{Os}$ .

2. (a) Following Fermi's theory of beta-decay, find out the probability of electron (beta) emission per unit time in the momentum range  $P_e$  and  $P_e + dP_e$ . (7)

(b) Show that the slope of the electron energy spectrum for allowed decays is zero near  $T_e = Q$  if  $m_\nu = 0$  but becomes infinite if  $m_\nu \neq 0$ . (3)

3. (a) What is Mossbauer effect? Write down the experimental evidence of Mossbauer effect. (2+1)

(b) Write down the selection rules for gamma-emission. (2)

(c) An even-Z, even-N nucleus has the following sequence of levels above its  $0^+$  ground state:  $2^+$  (89 keV),  $4^+$  (288 keV),  $6^+$  (585 keV),  $0^+$  (1050 keV),  $2^+$  (1129 keV). Draw an energy level diagram and show all reasonably probable  $\gamma$  transitions and their dominant multipole assignments. (5)

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**Internal Assessment-10**