2019

M.Sc.

1st Semester Examination

PHYSICS

PAPER - PHS-103 (Gr. - 103.1 + 103.2)

Full Marks: 50

Time: 2 Hours

Use separate answer scripts for 103.1 and Group 103.2

(Electrodynamics – PHS 103.1)

Answer Q1, Q2 and any one from Q3 and Q4

1. Answer any two bits:

2X2 = 4

- (a) Why the advance potential is not physically acceptable?
- (b) Define the distribution function in phase space under plasma kinetic theory.
- (c) Show that $F^{\mu\nu}G_{\mu\nu}=4\frac{\overrightarrow{E}.\overrightarrow{B}}{c}$, where $F^{\mu\nu}$ is field tensor and $G_{\mu\nu}$ is the dual.
- (d) Write difference between bremsstrahlung and cherenka radiation.
- 2. Answer any two bits:

2X4 = 8

- (a) Deduce Boltzmann equation for plasma.
- (b) Write down the Lienard-Wiechart potentials and hence calculate the magnetic field for a moving point charge.
- (c) Calculate the differential cross section for scattering by dipoles induced in small scatters.

(Turn Over)

Page - 02

- (d) Show that momentum four vector is constant using Hamiltonian action principle and $\partial_{\mu}F^{\mu\nu}=\mu_{0}J^{\nu}$ where $F^{\mu\nu}$ is field tensor, $G^{\mu\nu}$ it's dual and $J^{\nu}=(c\rho,\vec{J})$
- 3. Calculate the Potential and fields for an oscillating electric dipole and also calculate radiation energy for this case. (6+2)
- 4. (a) Write down the retarded potentials and hence deduce the time dependent generalization of Coulomb's law.
- (b) An infinite straight wire carries a linearly increasing current I(t) = Kt, t > 0; = 0, otherwise. Find the electric field generated. (5+3)

(Material Preparation & Characterization – PHS 103.2) Answer Q1, Q2 and any one from Q3 and Q4

1. Answer any two bits:

2X2 = 4

- (a) Why nanomaterials are more reactive than their bulk counterpart?
- (b) Why electron microscopy is better than the optical microscopy?
- (c) How will you distinguish between the XRD spectra of a crystalline and amorphous materials?
- (d) What is the significance of the peak observed in UV-VIS-NIR spectra?
- 2. Answer any two bits:

2X4 = 8

(a) State briefly the principle of melt-quench method to prepare glass and explain glass transition temperature.

(Continued)

- (b) Describe Atomic Layer Deposition (ALD) technique for the deposition of the Al₂O₃ thin film.
- (c) Give the principle of the optical absorption spectroscopy and explain with block diagram.
- (d) How the surface of a substrate can be cleaned? Considering very high kinetic energy (E) of an electron derive its de-Broglie wavelength in TEM. (2+2)
- 3. (a) Show schematically basic components of Atomic Force Microscopy (AFM) and explain the operational principle of the AFM?
- (b) Discuss briefly the different sections of Transmission Electron Microscope (TEM) and explain how it can be operated in "diffraction mode" and "image mode".
- (c) Name different pressure gauges with their range to measure the different levels of vacuum. (3+4+1)
- 4. (a) What is X-ray photoelectron spectroscopy? Write its working principle. What are the sources of X-ray used in this spectroscopy? What is the advantage of using synchrotron radiation as the source of X-ray? Describe the mechanisms of energy loss of electron in the X-ray photoelectron spectroscopy. (2+1+1+2+2)

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Internal Assessment-10